Thermal conductivity in intermetallic clathrates: A first-principles perspective

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Inorganic clathrates such as $Ba_8Ga_xGe_{46-x}$ and $Ba_8Al_xSi_{46-x}$ commonly exhibit very low thermal conductivities. A quantitative computational description of this important property has proven difficult, in part due to the large unit cell, the role of disorder, and the fact that both electronic carriers and phonons contribute to transport. Here, we conduct a systematic analysis of the temperature and composition dependence of low-frequency modes associated with guest species in $Ba_8Ga_xGe_{46-x}$ and $Ba_8Al_xSi_{46-x}$ ("rattler modes"), as well as thermal transport in stoichiometric $Ba_8Ga_{16}Ge_{30}$. To this end, we account for phonon-phonon interactions by means of temperature-dependent effective interatomic force constants, which we find to be crucial in order to achieve an accurate description of the lattice part of the thermal conductivity. While the analysis of the thermal conductivity is often largely focused on the rattler modes, here it is shown that at room temperatures modes with $\hbar \omega \gtrsim 10 \text{ meV}$ account for 50% of lattice heat transport. Finally, the electronic contribution to the thermal conductivity is computed, which shows the Wiedemann-Franz law to be only approximately fulfilled. As a result, it is crucial to employ the correct prefactor when separating electronic and lattice contributions for experimental data.

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I. INTRODUCTION

Thermoelectric materials enable the extraction of electrical power from a thermal gradient, as well as the reverse process: cooling through electrical power [1,2]. As a result, these materials are interesting for applications such as power generation in remote locations, waste heat recuperation, and active cooling. In the high-temperature region, which is of interest for waste heat recuperation from combustion processes, inorganic clathrates are among the most efficient thermoelectric materials [3,4] with studies reporting figure-of-merit (zT) values above 1 [5,6].

Clathrates are chemical substances with a defined lattice structure that can trap atomic or molecular species [7,8]. For thermoelectric applications, one usually considers inorganic clathrates, for example Ba₈Ga₁₆Ge₃₀ and Sr₈Ga₁₆Sn₃₀ [9,10]. Here, the earth alkaline atoms act as guest species that occupy the cages provided by the host structure, where the latter is most commonly composed of elements from groups 13 and 14. In the present paper, we focus on Ba₈Ga₁₆Ge₃₀, which belongs to space group $Pm\bar{3}n$ (International Tables of Crystallography number 223) and features two smaller and six larger cages per unit cell (Fig. 1) [9–11]. Ba₈Ga₁₆Ge₃₀ has been investigated extensively both experimentally [5,6, 12–16] and theoretically [13,17–24], especially because of its promising thermoelectric properties.

Generally, these materials exhibit a very low thermal conductivity, which is comparable to that of glasses [3,16,25]. This is crucial for their good thermoelectric performance and, accordingly, clathrates are commonly regarded as realizations of the "phonon glass–electron crystal" concept [26]. This behavior can be attributed to the "rattler"-like atomic motion of the guest species, which results from their relatively small size compared to the host cage [20,27].

While experimental measurements agree with the general magnitude of the thermal conductivity in clathrates, there are some noticeable differences (illustrated for the case of Ba₈Ga₁₆Ge₃₀ in Fig. 5 below). These variations can originate from several factors, including sample preparation, thermal emission, and the general difficulties associated with measuring small thermal conductivities. Since one is often interested in separating out the contribution to the thermal conductivity from lattice vibrations, one must make assumptions with respect to the electronic contribution, which adds another source of uncertainty. In this situation, predictive modeling on the basis of first-principles calculations can not only provide very valuable insight but also guide the development of new materials. Compared to previous studies on Ba₈Ga₁₆Ge₃₀ [23,24], we provide a comprehensive comparison with experimental data for both vibrational spectrum and thermal conductivity, as well as an analysis of the composition dependence of the rattler modes and the electronic contribution to the thermal conductivity. We furthermore emphasize the importance of accounting for the temperature dependence of phonon frequencies in order to obtain a sensible description of the thermal conductivity in this type of materials.

Here, we present a comprehensive study of the thermal conductivity in $Ba_8Ga_{16}Ge_{30}$ as a prototypical clathrate, which combines Boltzmann transport theory with firstprinciples calculations within the framework of density functional theory. We separately address the electronic and vibrational contributions, account for finite-temperature effects on vibrational frequencies and lifetimes, consider the impact of the exchange-correlation functional, and conduct a careful comparison with experimental data. It is demonstrated that the

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FIG. 1. Crystal structure of type I clathrates. The guest species (Ba) occupies Wyckoff sites of type 2a and 6d, while the host species (Ga, Ge) occupy Wyckoff sites of type 6c, 16i, and 24k.

Wiedemann-Franz law only provides a reasonable approximation to the electronic thermal conductivity if an appropriate prefactor is used; an aspect that is often overlooked in the analysis of experimental data.

Furthermore, it is shown that the rattler-mode frequencies calculated in the static (zero temperature) limit systematically underestimate the experimental data, which leads to a striking underestimation of the lattice thermal conductivity. To overcome this limitation, one must take into account phonon renormalization, which is accomplished effectively by using temperature-dependent force constants. The resulting model reproduces the experimentally measured temperature dependence of the rattler-mode frequencies and leads to an increase of the thermal conductivity by more than a factor of 2, which brings the predicted data into good agreement with experimental data. Finally, we provide a careful examination of the contributions to the thermal conductivity, which shines light on the "phonon glass" picture and reveals that more modes contribute to the thermal conductivity than previously assumed.

The remainder of this paper is organized as follows: The next section provides an overview of the methodological aspects of this work including computational details as well as a review of the most important relations from Boltzmann transport theory. The first part of the Results and Discussion section then focuses on the suitability of different exchange-correlation functionals to describe the clathrate structures. This is followed by an examination of the phonon dispersion, especially the rattler modes, emphasizing their temperature dependence as well as the effect of order and composition. This sets up a comprehensive analysis of the vibrational and electronic contributions to the thermal conductivity. Finally, we summarize the key results and conclusions in Sec. IV.

II. METHODOLOGY

A. Thermal conductivity

The thermal conductivity κ in a solid, chiefly, comprises contributions from electronic carriers κ_e and lattice vibrations (phonons) κ_l ,

$$\kappa = \kappa_e + \kappa_l. \tag{1}$$

In the following, we outline the approach taken to compute the two terms in the above equation.

1. Lattice thermal conductivity

The lattice (phononic) contribution κ_l to the thermal conductivity can be computed by solving the phonon Boltzmann transport equation (BTE) [28]. In the present treatment, we limit our analysis to the framework of the relaxation time approximation (RTA) of the phonon BTE, in which the lattice thermal conductivity is

$$\kappa_l = \frac{1}{\Omega} \sum_{iq} g_q \boldsymbol{v}_{iq} \otimes \boldsymbol{v}_{iq} \tau_{iq} c_{iq}.$$
(2)

Here, Ω is the unit cell volume, g_q is the q-point weight, and $v_{iq} = \nabla_q \omega_{iq}$ is the group velocity of mode i at point qof the Brillouin zone while ω_{iq} is the mode frequency. Both phonon frequencies and group velocities can be derived from the second-order force constant matrix, which is given by the second derivative of the energy E with respect to the atomic displacements $u_{\alpha}(il)$ [28],

$$\Phi_{\alpha\beta}(il,i'l') = \frac{\partial^2 E}{\partial u_{\alpha}(il)\partial u_{\beta}(i'l')},\tag{3}$$

where α and β are Cartesian directions, *i* is the site index relative to the unit cell basis, and *l* an index enumerating the unit cells. From the force constant matrix one can readily compute the dynamical matrix at any momentum vector *q*,

$$D_{\alpha\beta}(jj',\boldsymbol{q}) = \frac{1}{\sqrt{m_j m_{j'}}} \sum_{l'} \Phi_{\alpha\beta}(j0,j'l') e^{i\boldsymbol{q}\cdot(r_{j'l'}-r_{j0})}, \quad (4)$$

where m_j is the atomic mass of the species occupying site *j*. Diagonalization of $D_{\alpha\beta}(jj', q)$ then yields normal modes and phonon frequencies ω_{iq} , from which the mode-specific heat capacity c_{iq} at temperature *T* can be obtained via

$$c_{iq} = k_B \frac{x^2 \exp x}{(1 - \exp x)^2} \quad \text{with} \quad x = \frac{\hbar \omega_{iq}}{k_B T}.$$
 (5)

The relaxation time τ_{iq} , or phonon lifetime, which appears in Eq. (2), comprises contributions from different scattering processes, including phonon-phonon interaction, isotope mass variation, boundary scattering, alloying, and disorder. Here, we consider phonon-phonon interaction and isotope mass variation. According to the most simple approximation, known as Matthiessen's rule, the different scattering processes are assumed to be independent; i.e., their scattering rates (or inverse lifetimes) are additive:

$$\tau_{iq}^{-1} = \tau_{\text{ph-ph},iq}^{-1} + \tau_{\text{iso},iq}^{-1}.$$
 (6)

In the present work, scattering due to isotope mass variation $(\tau_{iso,iq})$ has been treated according to second-order perturbation theory [29] whereas the contribution due to phonon-phonon scattering $(\tau_{ph-ph,iq}^{-1})$ was treated at the level of first-order perturbation theory [28], which requires knowledge of not only the second- but also third-order interatomic force constants (IFCs) [30,31]. As detailed in Sec. II B, we computed IFCs both in the static (0 K) limit using the finite-displacement method and from molecular dynamics (MD) simulations.

2. Electronic thermal conductivity

In the relaxation time approximation (RTA) to the linearized Boltzmann transport equation (BTE), the electronic contribution to the thermal conductivity κ_e is given by

$$\boldsymbol{\kappa}_e = \boldsymbol{\kappa}^0 - \boldsymbol{S}^2 \boldsymbol{\sigma} T \tag{7}$$

with [28,32,33]

$$\boldsymbol{\sigma} = \frac{2e^2}{\Omega} \sum_{ik} g_k \boldsymbol{v}_{ik} \otimes \boldsymbol{v}_{ik} \tau_{ik} \left(\frac{\partial f}{\partial \varepsilon}\right)_{\varepsilon = \varepsilon_{ik}},\tag{8}$$

$$\mathbf{S} = \frac{\boldsymbol{\sigma}^{-1}}{eT} \sum_{ik} g_k \boldsymbol{v}_{ik} \otimes \boldsymbol{v}_{ik} \tau_{ik} [\varepsilon_{ik} - \mu_e] \left(\frac{\partial f}{\partial \varepsilon}\right)_{\varepsilon = \varepsilon_{ik}}, \quad (9)$$

$$\boldsymbol{\kappa}^{0} = \frac{e}{T\Omega} \sum_{ik} g_{k} \boldsymbol{v}_{ik} \otimes \boldsymbol{v}_{ik} \tau_{ik} [\varepsilon_{ik} - \mu_{e}]^{2} \left(\frac{\partial f}{\partial \varepsilon}\right)_{\varepsilon = \varepsilon_{ik}}.$$
 (10)

Here, Ω is the unit cell volume, g_k is the *k*-point weight, *i* refers to the band index, τ_{ik} is the mode- and momentumdependent lifetime, $v_{ik} = \hbar^{-1} \partial \varepsilon_{ik} / \partial k$ is the group velocity, *f* is the occupation function, and μ_e is the electron chemical potential.

We have previously studied the electronic conductivity σ and the Seebeck coefficient **S** for Ba₈Ga₁₆Ge₃₀ and conducted a systematic comparison with experiment [21]. Using a charge carrier concentration of $n_e = 3 \times 10^{20}$ cm³ and a mode- and momentum-independent effective lifetime model with $\tau_{eff} =$ $\tau_{300}(300 \text{ K/T})^{1/2}$ we were able to achieve very good agreement with experimental data, and, accordingly, this approach is also adopted in the present work.

B. Computational details

1. General

DFT calculations were performed using the projector augmented wave method [34] as implemented in the Vienna *ab initio* simulation package (VASP) [35]. To assess the sensitivity of our results to the treatment of exchange-correlation effects, we used both the PBE functional [36] and the van der Waals density functional method [37] with consistent exchange (vdW-DF-cx) [38] as implemented in VASP [39,40].

The plane-wave energy cutoff was set to 243 eV (Ba₈Ga₁₆Ge₃₀), 312 eV (Ba₈Al₁₆Ge₃₀), and 319 eV (Ba₈Al₁₆Si₃₀, Ba₈Ga₁₆Si₃₀) in calculations with fixed volume and cell shape. For cell shape relaxations, the plane-wave energy cutoff was increased by 30%. A Gaussian smearing with a width of 0.1 eV was used throughout. Structural relaxations were performed using a Γ -centered 3 × 3 × 3 *k*-point mesh until the residual forces were below 10 meV/Å and absolute stresses were below 0.1 kbar.

2. Vibrational spectra and lattice thermal conductivity

The static second- and third-order IFCs as well as the thermal conductivity were computed using the SHENGBTE [30,41,42] and PHONOPY [43] codes. Calculations were carried out using $2 \times 2 \times 2$ supercells (432 atoms) and, in the case of the third-order IFCs, included displacements up to the fifth-neighbor shell. The Brillouin zone was sampled using a $9 \times 9 \times 9 q$ -point mesh and a smearing parameter of $\sigma = 0.01$. There was no indication of any significant difference

between the results within the framework of RTA-BTE and the fully converged solution to the BTE; hence, the full set of computations was limited to the RTA. The second-order IFCs obtained in this process were also used to model the thermal expansion within the quasiharmonic approximation (QHA).

As will be shown below, using the IFCs obtained in the static (0 K) limit to predict the thermal conductivity leads to a substantial underestimation. We, therefore, determined effective temperature-dependent IFCs using an approach similar to the one described in Ref. [44]. To this end, we carried out first-principles molecular dynamics (MD) simulations in the canonical ensemble at temperatures of 100, 200, 300, and 600 K. We employed primitive 54-atom cells, which were sampled using a Γ -centered 3 \times 3 \times 3 k-point mesh. The equations of motion were integrated for a total of about 5500 time steps using a step size of 5 fs. After discarding the first 1000 steps for equilibration, about 180 snapshots, selected at a spacing of 25 MD steps, were used for training, by least-squares fitting temperature-dependent interatomic force constants (TDIFCs) using our in-house HIPHIVE code [45]. Finally, SHENGBTE was used to calculate the thermal conductivity from the resulting IFCs.

3. Vibrational spectra of nonstoichiometric compounds

Additional calculations of the vibrational spectra were carried out for $Ba_8Ga_xGe_{46-x}$ and $Ba_8Al_xSi_{46-x}$, with $14 \le x \le 18$. To this end, we employed 54-atom cells and structures previously obtained through Monte Carlo simulations, which are representative of the actual chemical order in the material. In total, data were obtained for 20 structures per composition, equivalent to 200 configurations in total.

4. Electronic contribution to the thermal conductivity

To calculate the electronic contribution to the thermal conductivity, we considered both the chemically ordered ground state structure and the chemically disordered structures obtained by Monte Carlo simulations [21]. The latter configurations are representative of the actual chemical (dis)order in the material at 600, 900, and 1200 K, as is described in detail in Ref. [21]. Results were averaged over five structures per temperature. The wave functions of the fully relaxed structures were converged using a Γ -centered 4 × 4 × 4 *k*-point mesh, which was followed by a non-self-consistent computation of the eigenenergy spectra on a Γ -centered 20 × 20 × 20 mesh. The terms in Eq. (7) were, subsequently, computed using the BOLTZTRAP code [33].

III. RESULTS AND DISCUSSION

A. Structure and thermal expansion

An accurate description of the vibrational properties is important for modeling the thermal conductivity. While the PBE functional has been used extensively in the past for electronic structure calculations of inorganic clathrates (see, e.g., Refs. [20,23]), it is known to underestimate the bond stiffness (see, e.g., Ref. [46]). For this reason, we also considered the vdW-DF-cx method (see Sec. II B), which has been found to yield an excellent description of finite-temperature properties for other materials [46,47].

TABLE I. Finite-temperature lattice parameters from calculation and experiment. The values in parentheses denote the lattice parameters obtained without taking into account zero-point vibrations. In the calculations the temperature dependence was described at the level of the quasiharmonic approximation.

Material	Calculation		Experiment	
	0 K	300 K	300 K	
Ba ₈ Ga ₁₆ Ge ₃₀				
PBE	10.98 (10.96)	11.02	10.76 ^a [52]	10.80 ^a [14]
vdW-DF-cx	10.86 (10.85)	10.90		
Ba8Ga16Si30				
PBE	10.68 (10.66)	10.71	10.54 ^b [11]	10.57 ^b [48]
vdW-DF-cx	10.60 (10.58)	10.62		
Ba ₈ Al ₁₆ Si ₃₀				
PBE	10.74 (10.72)	10.76	10.64 ^b [50]	10.65 ^b [49]
vdW-DF-cx	10.67 (10.65)	10.69		
Ba ₈ Al ₁₆ Ge ₃₀				
PBE	11.01 (10.99)	11.04	10.85 ^a [51]	10.88 ^b [53]
vdW-DF-cx	10.91 (10.90)	10.94		

^aInterpolated.

^bExtrapolated.

As a first step in assessing the performance of these functionals, we determined the temperature dependence of the lattice structure of the four stoichiometric compounds Ba₈Ga₁₆Ge₃₀, Ba₈Ga₁₆Si₃₀, Ba₈Al₁₆Ge₃₀, and Ba₈Al₁₆Si₃₀, based on the ordered ground state (0 K) structure identified in Ref. [22]. In this context, it must be noted that it is not always possible to find experimental data for the lattice parameters for the stoichiometric compounds. This particularly applies to the Si-based clathrates [11,48–50], presumably because the solubility limit is about x = 15 [48,49] for both Ga in $Ba_8Ga_xSi_{46-x}$ and Al in $Ba_8Al_xSi_{46-x}$. The lattice parameters that correspond to x = 16 have been estimated by performing a linear least-squares fit for each data set. One should also note that the experimentally determined lattice parameters and compositions can vary markedly depending on the measurement technique [11,48,51].

Both functionals overestimate the lattice parameter compared to experiment, with PBE always giving the higher estimate (Table I and Fig. 2). Overall, the agreement achieved by the vdW-DF-cx calculations is very good, with an average deviation of 0.6% (1.5% for PBE).

It should be noted that chemical disordering, which is generally present in these compounds [11], has an effect on the lattice parameter. Based on our earlier analysis [22], one can assume that as the material is cooled down after synthesis the chemical order is frozen in at a temperature of about 600 K. At this temperature the lattice parameter in Ba₈Ga₁₆Ge₃₀ has been predicted to be increased by 0.028 Å relative to the ground state structure, decreased by about 0.028 Å in the case of Ba₈Ga₁₆Si₃₀, and remain relatively unchanged for Ba₈Al₁₆Ge₃₀ and Ba₈Al₁₆Si₃₀. These contributions, however, barely affect the agreement with experiment and leave the average errors unchanged.



FIG. 2. Temperature dependence of the lattice parameter for $Ba_8Ga_{16}Ge_{30}$ obtained within the quasiharmonic approximation. Experimental data are from Ref. [11].

B. Phonon dispersion in the static (0 K) limit

Due to the large mass of the Ba atoms as well as their weak coupling to the host structure, the associated rattler modes show up as low-frequency optical modes in the phonon spectrum (Fig. 3). They appear at higher frequencies in the vdW-DF-cx calculations [Fig. 3(a)] than in the case of the PBE functional [Fig. 3(b)] as expected based on the known "softness" of the latter. For both functionals one observes the phonon modes in the zero-temperature limit to be lower than the experimentally measured frequencies [black dashed lines in Figs. 3(a)–3(c)] [27].

C. Phonon dispersion at finite temperatures

As will be shown below, the thermal conductivity calculated on the basis of the static IFCs systematically and substantially underestimates experimental data (also see Ref. [23]). As will become apparent in the analysis of the thermal conductivity (Sec. IIIE), lower frequencies of the rattler modes reduce the Brillouin zone volume corresponding to propagating modes, which translates into a lower thermal conductivity. Therefore, a very relevant question is to which extent phonon-phonon interactions affect the rattler-mode frequencies. In fact, the low frequencies of the rattler modes imply that they are fully activated already at low temperatures and, thus, phonon-phonon interaction driven frequency shifts can already occur below room temperature. This notion is supported by experimental data from both inelastic neutron and Raman scattering [27], which reveal a notable temperature dependence of the rattler modes.

Using a series of temperature-dependent interatomic force constants (TDIFCs; see Sec. II B), we therefore calculate the vibrational spectrum as a function of temperature. The full phonon dispersion at 300 K [Fig. 3(c)] does indeed reveal an upward shift for the lowermost optical branches. A comprehensive comparison with experimental data [Fig. 4(c)] demonstrates that the TDIFCs can also rather accurately reproduce both the absolute values and the temperature dependence of the rattler modes [61]. As will be discussed below, these effects are actually crucial for being able to predict correctly the thermal conductivity.



FIG. 3. (a)–(c) Phonon dispersion of $Ba_8Ga_{16}Ge_{30}$ showing the low-frequency region along the Γ -R direction derived (a), (b) from IFCs obtained in the static (0 K) limit by the finite-displacement (FD) method and (c) from temperature-dependent interatomic force constants (TDIFCs) corresponding to a temperature of 300 K. Black dashed lines indicate the result of a simple spring model fitted to experimental data [27]. The model underestimates the splitting at the zone boundary close to 5 meV, which is actually approximately 1 meV. (d) Total phonon densities of states. (e) Partial densities of states showing the contributions from Ba on 2a Wyckoff sites (blue line), Ba on 6d sites (red line), and contributions from the Ga/Ge cage structure (green).

D. Chemical composition and ordering

According to experimental [49–51,57,62–67]; theoretical [22,68]; and combined [53,69] studies, the structural and physical properties of ternary, inorganic clathrates vary markedly with chemical composition. Specifically, it has been shown that the displacement of the guest atom from the 6*d* site [22,51] and the associated vibrational frequencies [57,62] depend on the number of Al or Ga atoms per unit cell, in ternary compounds of the type Ba₈{Al, Ga}_{*x*}{Si, Ge}_{46-*x*}. Moreover, experimental evidence suggests that the degree of off-centering, the frequencies of the lowest Raman-active modes, and the lattice thermal conductivity are correlated for compounds in the structurally similar quaternary system Sr₈Ga₁₆Si_{*x*}Ge_{30-*x*} [55,56]. Given these results, it is reasonable to assume that the lattice thermal conductivity also varies with chemical composition, to some degree. We therefore computed the variation of the 18 lowest-frequency phonon modes, associated with the Ba atom at the 6*d* site [70], with the number of group-13 atoms not only in Ba₈Ga_xGe_{46-x} but also Ba₈Al_xSi_{46-x}, since the latter represents the limit of a host matrix made up of light elements. Specifically, we extracted and averaged the Γ -point frequencies for 20 representative configurations (Sec. II B 3) for each composition in the range $14 \le x \le 18$. The modes naturally fall into three groups, with six modes in each [Figs. 4(a) and 4(b)]. The splitting of the modes can be viewed as a consequence of the following facts: (i) the guest atom is not located at the immediate center of the cage, (ii) the latter is shaped like a tetrakaidecahedron, and (iii) the



FIG. 4. (a), (b) Composition dependence of the lowest-lying rattler modes in (a) $Ba_8Ga_xGe_{46-x}$ and (b) $Ba_8Al_xSi_{46-x}$ along with the variation of the lattice constant (in gray). Specifically, the red, yellow, and blue lines correspond to the average frequencies for modes 1–6, 7–12, and 13–18, respectively. The standard deviations, obtained by averaging over 20 representative configurations at each composition, are indicated by shaded filled curves. (c) Temperature dependence of the partial density of states associated with Ba atoms on 6d Wyckoff sites [shaded filled curves; compare Fig. 3(e)] in comparison with experimental data from inelastic neutron scattering on powder (PINS) and single crystalline (SINS) samples [27], Raman scattering [27,54–57], and THz spectroscopy [58–60].

Al and Ga atoms are not necessarily symmetrically distributed between the framework sites [22].

For Ba₈Al_xSi_{46-x} the phonon modes slightly soften with increasing x; a similar trend, albeit even weaker, can also be observed for the lower two groups in the case of Ba₈Ga₁₆Ge₃₀. This behavior correlates with the increase in the lattice constant, which is larger for Ba₈Al_xSi_{46-x} than for Ba₈Ga_xGe_{46-x} [gray lines in Figs. 4(a) and 4(b)]. A larger lattice parameter implies that the size of the cages occupied by Ba atoms increases, which leads to weaker restoring forces and, hence, lower vibrational frequencies. One must conclude, however, that the phonon frequencies, associated with the vibrations of the Ba atom at the 6*d* Wyckoff, are relatively insensitive to the chemical composition since the difference between x = 14and x = 18 is of the same magnitude as the spread of the frequencies.

E. Thermal conductivity: Lattice contribution

1. Comparison of scattering channels

Based on the analysis of the vibrational properties described in the previous sections we computed the lattice contribution to the thermal conductivity. These calculations were carried out for the chemically ordered ground state structure of $Ba_8Ga_{16}Ge_{30}$ [21,22] using both the PBE functional and the vdW-DF-cx method.

When limiting the analysis to phonon-phonon scattering, one obtains a strong variation with temperature that follows a T^{-2} behavior at low temperatures and a T^{-1} trend at temperatures $\gtrsim 100 \text{ K}$ [Fig. 5(a)]. Isotope scattering affects only the very low temperature region leading to a peak in the conductivity below 10 K.

The chemical disordering at finite temperatures that was already alluded to above is inevitably associated with mass mixing, which affects both the frequencies via Eq. (4) and the lifetimes in a way analogous to isotope scattering. The effect on the frequencies is modest as evident from the analysis in Sec. III D; that is to say, the second-order force constants are hardly affected by chemical mixing on the host lattice, at least at a Ga : Ge ratio of 16 : 30.

By extension of Eq. (6), mass disorder due to chemical mixing can, however, also affect the lifetimes. Here, this effect was modeled analogously to isotope (mass) scattering using site occupancy factors obtained in previous simulations to compute the variance of the atomic masses [22]. The results show that the impact on the lifetimes is only discernible at very low temperatures [dashed black line in Fig. 5(a)]. This is in fact rather unsurprising since elemental Ge has a relatively broad isotope distribution.

2. Overview of experimental data

The total thermal conductivity κ has been measured using a variety of techniques for both single and polycrystalline samples; see, e.g., Refs. [5,6,63,72]. To extract the lattice contribution to the thermal conductivity, it is customary to remove the electronic part by the use of the Wiedemann-Franz law. Yet, as discussed in Sec. III F, the Wiedemann-Franz law is in fact only approximately valid under the relevant conditions. For the sake of consistency, here we nonetheless



FIG. 5. Lattice thermal conductivity κ_l of Ba₈Ga₁₆Ge₃₀ as a function of temperature. (a) Comparison of κ_l due to different included scattering channels, calculated using IFCs achieved with the vdW-DF-cx functional, as well as a comparison to the computations using temperature-dependent interatomic force constants (TDIFCs). (b) Comparison between computations, using IFCs from the vdW-DF-cx functional (red solid line), the PBE functional (solid blue line), temperature-dependent force constants (squares), and experimental data sets (dashed lines marked by numbers). The inset shows the same data on a linear scale. The experimental data are from Refs. [71] (1, *p* type), [64] (2, *p* type; 3, *n* type), [12] (4, *n* type), [72] (5, n type), and [73] (6, 7). The authors of Ref. [72] (5) point out that due to the large surface to volume ratio, their measurements become unreliable above approximately 100 K due to thermal emission. The shaded areas and open squares represent the difference between the electronic contribution to the thermal conductivity from BTT κ_e and the Wiedemann-Franz law $L\sigma T$ with $L = 2.0(k_B/e)^2$.

use the data reported in the original papers and focus on single-crystalline samples.

The compilation of the experimental results [Fig. 5(b)] illustrates a noticeable spread, especially at low temperatures. The strong temperature dependence along with the pronounced low-temperature peak in the data for *n*type $Ba_8Ga_{16}Ge_{30}$ [63,72] indicate a crystal-like thermal conductivity down to very low temperatures. By contrast, the much weaker temperature dependence and low-temperature plateau in the results for *p*-type material [63,71] are consistent with glass-like behavior.

This correlation between *n*-type (*p*-type) electrical and crystal-like (glass-like) thermal conductivity has been documented in Ref. [16]. The crossover between the two conductivity types occurs close to the stoichiometric composition of 16 : 30 with *n*-type (*p*-type) material being slightly Ga (Ge) deficient [63]. It has been found experimentally [14] that Ba atoms, which are formally assigned to 6*d* Wyckoff sites, are less (more) displaced in *n*-type (*p*-type) material [22].

Interestingly, a numerical study of the low-frequency modes in Ba8Ga16Ge30 and Ba8Ga16Sn30, as cases for on- and off-center Ba positioning, has found crystal-like and glass-like conductivity, respectively [74]. For the case of $Ba_8Ga_{16}Sn_{30}$, it has, furthermore, been argued that the low-temperature plateau is due to a delocalization-localization transition for the acoustic modes [75]. The latter process can, in turn, only occur for off-center guest atoms, which induce disorder and level repulsion. In this situation, the heat-carrying quasiparticles become overdamped; i.e., the oscillation period is comparable to the lifetime, leading to a saturated and, thus, temperatureindependent lattice thermal conductivity [76]. To properly capture this effect, a nonperturbative treatment is required [77]. Here, we therefore limit ourselves to a comparison with experimental data for the thermal conductivity of *n*-type material.

3. Comparison between calculations and experiment

The lattice thermal conductivities derived from IFCs obtained in the static limit by either PBE or vdW-DF-cx substantially underestimate the experimental data (Fig. 5; also see Ref. [23]). This is a rather unusual observation as one more commonly finds calculations to overestimate the experimental data (see, e.g., Ref. [47]), as computational analysis commonly account only for some of the scattering mechanisms that are active in reality.

The behavior observed here can, however, be understood by considering the expression for the lattice thermal conductivity Eq. (2) and the phonon dispersions (Fig. 3). In the case of the static IFC calculations, the rattler modes are located at lower frequencies than in the experimental data and, consequently, the avoided crossings with the acoustic modes [27] occur at smaller q vectors. As a result, the relative fraction of propagating modes that contributes to the thermal conductivity is reduced. This mechanism can also explain the lower thermal conductivity obtained from PBE relative to vdW-DF-cx calculations.

As shown above, one must account for the effect of phonon-phonon coupling on the frequency spectrum in order to obtain closer agreement with experiments (Fig. 3). Accordingly, using temperature-dependent IFCs (TDIFCs) in Eq. (2) leads to a substantial increase in the thermal conductivity, more precisely by a factor of 2 at 100 K up to a factor of 3 at 600 K. This effect is crucial in order to achieve good agreement with the experimental data in this temperature range. In addition, the temperature dependence of κ_l shifts from $\kappa_l \propto T^{-1}$ to $\kappa_l \propto T^{-0.69}$, which rather closely follows the experimental trend.

The importance of temperature-dependent frequencies is also evident from the calculations in Ref. [24], in which the temperature dependence of the vibrational spectrum was included via the self-consistent phonon formalism based on a fourth-order model. In the latter approach, the frequencies are temperature dependent whereas the IFCs themselves are not. In the present case, we employ (effective) TDIFCs up to third order to achieve a very similar effect [78]. The present methodology moreover allows us to analyze the contribution from second- and third-order TDIFCs to the thermal conductivity separately. To this end, we also computed the thermal conductivity using second-order TDIFCs and static (0 K) third-order IFCs. The results (see Fig. 1 of the Supplemental Material [79]) demonstrate the (effective) temperature dependence to be important for both second- and third-order IFCs. Whereas including (only) second-order TDIFCs causes a change in the structure of the lifetimes as a function of frequency, including third-order TDIFCs, additionally, leads to a systematic increase in the lifetimes by a factor of up to approximately 2.

4. Microscopic contributions

Inorganic clathrates have repeatedly been shown to exhibit very low thermal conductivities [3,5,6,63,72] and have been discussed as realizations of the so-called phonon glass–electron crystal (PGEC) concept [1,16]. Accordingly, the mechanisms that give rise to the very low thermal conductivity in clathrates have been scrutinized experimentally as well as by theory and simulations; see, e.g., Refs. [20,23-25,27,74, 80-83]. The primary object of attention has been the phonon dispersion [20,27,74], which provides information about vibrational frequencies and group velocities and is more readily accessible both in experiments and calculations. The phonon lifetimes in these materials have only been recently addressed, using an approximate lifetime model for Ba₈Ga₁₆Si₃₀ [80] as well as perturbation theory in the case Ba₈Ga₁₆Ge₃₀ [23,23,24].

The present calculations allow us to scrutinize the individual contributions to the lattice thermal conductivity according to Eq. (2), including group velocities and lifetimes, and, thereby, gain further insight into the glass-like thermal conduction in inorganic clathrates. From the phonon dispersion [Fig. 6(a)], two energy regions, specifically $\omega < 3 \text{ meV}$ and 12 meV $< \hbar \omega < 16$ meV, can be identified with large group velocities. At low temperatures, the contribution to κ_L stems mainly from the region $\omega < 2 \text{ meV}$ [Fig. 6(b)]. Yet already at about 300 K almost half of the heat transport is accomplished by modes with frequencies above 3 meV. This behavior is the result of two concurrent processes: (i) from about 300 K all modes are occupied and the mode specific heat capacity Eq. (5) saturates; (ii) the lifetimes of the acoustic modes in the lower energy window drop substantially whereas the lifetimes of the higherlying modes are relatively less affected [Fig. 6(d)]. Compared to previous studies, which focused primarily on the low-energy region, the present results demonstrate that, at least in the case of Ba8Ga16Ge30, important contributions stem from higher-energy modes and that these must be



FIG. 6. (a) Phonon dispersion in $Ba_8Ga_{16}Ge_{30}$ from static vdW-DF-cx IFCs. (b) Normalized accumulated lattice thermal conductivity, with respect to energy, as computed with static vdW-DF-cx IFCs at 3, 30, and 300 K (solid lines), compared to calculations based on the temperature-dependent interatomic force constants (TDIFCs) at 300 K. (c) Phonon-phonon limited lifetimes at 300 K computed with PBE IFCs (black markers), vdW-DF-cx IFCs (red markers), and TDIFCs (blue markers). The filled yellow curves indicate the overdamped region for a classical harmonic oscillator. (d) Comparison of lifetimes computed from static vdW-DF-cx IFCs at 3, 30, and 300 K.

included in order to obtain a sound description of the thermal conductivity.

When considering the static IFCs, it appears that a large number of modes is, actually, either strongly damped or even overdamped [Fig. 6(c)]. The oscillation period of these modes is thus comparable to their lifetime. In the case of the TDIFCs, the lifetimes are notably longer, which reflects the effective (albeit not formal) renormalization of the modes.

In agreement with previous work, the present calculations demonstrate that the weak coupling between host and lowlying guest (Ba) modes gives rise to avoided band crossings in the phonon dispersion [20,27]. In turn this causes a dramatic reduction in the group velocities of almost all modes with frequencies above the lowest guest mode [20,27]. The few dispersed modes above this threshold are strongly damped and, accordingly, do not contribute notably to κ_l . The very small thermal conductivity is thus the result of the extremely small Brillouin zone volume available to propagating phonon modes. Since the effective mean-free path drops to about 1 nm already at 300 K, further reduction of κ_l by, e.g., microstructural engineering [84] does not appear to be very promising. It should also be recalled that in the temperature range relevant for thermoelectric applications, the electronic contribution κ_e is already comparable to, if not larger than, κ_l (see Sec. III F).

F. Thermal conductivity: Electronic contribution

When conducting experiments, only the total thermal conductivity κ is directly accessible. To resolve each contribution, one therefore commonly resorts to the Wiedemann-Franz law. The Wiedemann-Franz law couples the electrical conductivity σ to the electronic thermal conductivity

$$\kappa_e = L\sigma T,\tag{11}$$

by which the lattice thermal conductivity κ_l is estimated. Here, $L = L_0(k_B^2/e^2)$ is the Lorenz number. When resolving experimental data, it is common to use $L_0 = \pi^2/3 \approx 3.3$, which is obtained for a degenerate electron gas, or not specify the value of L_0 used. As noted, e.g., in Ref. [85], a value of $L_0 = 2$ should be used for a degenerate semiconductor (such as a typical thermoelectric clathrate). Even then, the Wiedemann-Franz law ought to be considered a low-level approximation to the actual behavior as L_0 is not a universal constant.

We therefore conclude our analysis of the thermal conductivity in the prototypical inorganic clathrate Ba₈Ga₁₆Ge₃₀ by considering the electronic contribution κ_e (Fig. 7). At low temperatures (≤ 200 K) the thus obtained κ_e agrees rather well with the Wiedemann-Franz law for a degenerate electron gas ($L_0 \approx 3.3$), whereas at higher temperatures (≥ 400 K) the semiconductor value $L_0 = 2$ yields better agreement.

Above approximately 900 K, the κ_{e} from Boltzmann transport theory reveals a sharp rise for some structural models, which is due the onset of bipolar conduction, caused by the activation of minority charge carriers [5]. In this context, one should note that the integrand in the expression for the electronic thermal conductivity [33,86] Eq. (10) includes a term $(\varepsilon_{ik} - \mu_e)^2$. As a result, κ_e is most sensitive to contributions from states about k_BT above and below the Fermi level μ_e , rather than to states in the immediate vicinity of μ_e , which dominate in the case of σ . This is one of the shortcomings of the effective lifetime model, which does not distinguish these states. In any case, the present analysis suggests that the Wiedemann-Franz law should be applied with caution when trying to discriminate the electronic and lattice thermal conductivities, and that a value of L_0 corresponding to a degenerate semiconductor is more appropriate for describing the situation in inorganic clathrates.



FIG. 7. Electronic contribution κ_e to the thermal conductivity calculated for the ground state structure as well as for structures extracted from Monte Carlo simulations [21] representative of the chemical order at different temperatures. Data obtained using the Wiedemann-Franz law $\kappa_e = L\sigma T$ are shown by dashed lines, whereas the thermal conductivity obtained within the framework of Boltzmann transport theory [33,86] is shown by solid lines. Note that below 900 K the two sets of data deviate by as much as 25%, whereas above approximately 900 K the BTT data indicate a sharp rise that is not predicted by the Wiedemann-Franz law.

IV. CONCLUSIONS

The very low thermal conductivities observed in inorganic clathrates are challenging to address both experimentally and computationally. In the present study, focusing on $Ba_8Ga_{16}Ge_{30}$, we have undertaken a systematic computational analysis of the various mechanisms and features that contribute to this property.

First, we have addressed the role of the exchangecorrelation functional in describing both structure and vibrational spectra, from which we concluded that the vdW-DF-cx method provides a well balanced description of inorganic clathrates. Next, by using temperature-dependent interatomic force constants we demonstrated that phonon-phonon coupling (and thus temperature) must be taken into account in order to accurately capture the frequencies of the rattler modes, which also allowed us to correctly predict the experimentally observed temperature dependence of these modes. On the other hand, the composition dependence of the rattler-mode frequencies was shown to be small.

Based on this level of understanding, we then predicted both the lattice and the electronic thermal conductivity. For the former, we obtained very good agreement with experiments using temperature-dependent IFCs whereas we observed a pronounced underestimation when using IFCs representing the static limit. The κ_l values obtained using different IFCs could be rationalized by considering the relative Brillouin zone volume of propagating (heat carrying) modes. Specifically, the underestimation of the rattler modes is associated with the onset of avoided crossings at a lower q vector. The analysis, furthermore, reveals that IFCs obtained in the static limit yield heavily damped as well as overdamped quasiparticles, which is suggestive of glass-like transport. Taking into account phonon-phonon coupling via temperature-dependent IFCs (and thus effectively mimicking renormalization) leads to larger lifetimes and more well defined quasiparticles.

Finally, a comparison of predictions for the electronic contribution to the thermal conductivity demonstrates that the Wiedemann-Franz law must be applied with care when separating experimental thermal conductivity data, as the L_0 prefactor can vary between 2 (degenerate semiconductor) and 3.3 (free-electron gas) across the temperature range of interest.

The present results provide a very detailed perspective on the thermal conductivity in inorganic clathrates and shed light on the application of Boltzmann transport theory for predicting systems with strongly damped quasiparticles. We recognize that there exist a few, computational, studies on the thermal conductivity in clathrates, including Ba₈Ga₁₆Ge₃₀ [23,24], pure silicon clathrates $M_{8-x}Si_{46}$ [81,83], and the ordered clathrate Ba₈Au₆Ge₄₆ [82]. With the exception of Ref. [24], however, these studies employed IFCs obtained at 0 K and restricted the range of interactions. Here, we emphasized the importance of taking into account the effect of phonon-phonon coupling on the vibrational spectrum. Furthermore, we demonstrated the efficacy of advanced regression schemes [45] for extracting IFCs with minimal restrictions.

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- D. Rowe (Ed.), *Thermoelectrics Handbook: Macro to Nano* (CRC Press, Boca Raton, FL, 2005).
- [2] G. J. Snyder and E. S. Toberer, Nat. Mater. 7, 105 (2008).
- [3] J. L. Cohn, G. S. Nolas, V. Fessatidis, T. H. Metcalf, and G. A. Slack, Phys. Rev. Lett. 82, 779 (1999).
- [4] B. B. Iversen, A. E. Palmqvist, D. E. Cox, G. S. Nolas, G. D. Stucky, N. P. Blake, and H. Metiu, J. Solid State Chem. 149, 455 (2000).
- [5] A. Saramat, G. Svensson, A. E. C. Palmqvist, C. Stiewe, E. Mueller, D. Platzek, S. G. K. Williams, D. M. Rowe, J. D. Bryan, and G. D. Stucky, J. Appl. Phys. 99, 023708 (2006).
- [6] E. S. Toberer, M. Christensen, B. B. Iversen, and G. J. Snyder, Phys. Rev. B 77, 075203 (2008).
- [7] A. D. McNaught and A. Wilkinson, *IUPAC: Compendium of Chemical Terminology*, 2nd ed. (Blackwell Scientific Publications, Oxford, 1997) [XML online corrected version:

http://goldbook.iupac.org (2006), created by M. Nic, J. Jirat, and B. Kosata, updates compiled by A. Jenkins].

- [8] G. P. Moss, P. A. S. Smith, and D. Tavernier, Pure Appl. Chem. 67, 1307 (2009).
- [9] P. Rogl, in *Thermoelectrics Handbook*, edited by D. Rowe (CRC Press, Boca Raton, FL, 2005), Chap. 32, pp. 1–24.
- [10] A. V. Shevelkov and K. Kovnir, in *Zintl Phases*, Structure and Bonding No. 139, edited by T. F. Fässler (Springer, Berlin, 2011), pp. 97–142.
- [11] M. Christensen, S. Johnsen, and B. B. Iversen, Dalton Trans. 39, 978 (2010).
- [12] B. C. Sales, B. C. Chakoumakos, R. Jin, J. R. Thompson, and D. Mandrus, Phys. Rev. B 63, 245113 (2001).
- [13] J. D. Bryan, N. P. Blake, H. Metiu, G. D. Stucky, B. B. Iversen, R. D. Poulsen, and A. Bentien, J. Appl. Phys. 92, 7281 (2002).
- [14] M. Christensen, N. Lock, J. Overgaard, and B. B. Iversen, J. Am. Chem. Soc. 128, 15657 (2006).
- [15] D. Cederkrantz, A. Saramat, G. J. Snyder, and A. E. C. Palmqvist, J. Appl. Phys. **106**, 074509 (2009).
- [16] T. Takabatake, K. Suekuni, T. Nakayama, and E. Kaneshita, Rev. Mod. Phys. 86, 669 (2014).
- [17] N. P. Blake, S. Latturner, J. D. Bryan, G. D. Stucky, and H. Metiu, J. Chem. Phys. 115, 8060 (2001).
- [18] N. P. Blake, D. Bryan, S. Latturner, L. Møllnitz, G. D. Stucky, and H. Metiu, J. Chem. Phys. **114**, 10063 (2001).
- [19] G. K. H. Madsen, K. Schwarz, P. Blaha, and D. J. Singh, Phys. Rev. B 68, 125212 (2003).
- [20] G. K. H. Madsen and G. Santi, Phys. Rev. B 72, 220301(R) (2005).
- [21] M. Ångqvist, D. O. Lindroth, and P. Erhart, Chem. Mater. 28, 6877 (2016).
- [22] M. Ångqvist and P. Erhart, Chem. Mater. 29, 7554 (2017).
- [23] T. Tadano, Y. Gohda, and S. Tsuneyuki, Phys. Rev. Lett. 114, 095501 (2015).
- [24] T. Tadano and S. Tsuneyuki, Phys. Rev. Lett. 120, 105901 (2018).
- [25] J. Dong, O. F. Sankey, and C. W. Myles, Phys. Rev. Lett. 86, 2361 (2001).
- [26] D. Rowe (eds.), CRC Handbook of Thermoelectrics (CRC Press, Boca Raton, FL, 1995).
- [27] M. Christensen, A. B. Abrahamsen, N. B. Christensen, F. Juranyi, N. H. Andersen, K. Lefmann, J. Andreasson, C. R. H. Bahl, and B. B. Iversen, Nat. Mater. 7, 811 (2008).
- [28] J. M. Ziman, *Electrons and Phonons* (Clarendon Press, Oxford, 1960).
- [29] S.-i. Tamura, Phys. Rev. B 27, 858 (1983).
- [30] W. Li, J. Carrete, N. A. Katcho, and N. Mingo, Comput. Phys. Commun. 185, 1747 (2014).
- [31] W. Li, J. Carrete, and N. Mingo, Appl. Phys. Lett. 103, 253103 (2013).
- [32] N. P. Blake, L. Møllnitz, G. Kresse, and H. Metiu, J. Chem. Phys. 111, 3133 (1999).
- [33] G. K. Madsen and D. J. Singh, Comput. Phys. Commun. 175, 67 (2006).
- [34] P. E. Blöchl, Phys. Rev. B 50, 17953 (1994); G. Kresse and D. Joubert, *ibid.* 59, 1758 (1999).
- [35] G. Kresse and J. Furthmüller, Comput. Mater. Sci. 6, 15 (1996).
- [36] J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. 77, 3865 (1996); 78, 1396(E) (1997).

- PHYSICAL REVIEW B 100, 045206 (2019)
- [37] K. Berland, V. R. Cooper, K. Lee, E. Schröder, T. Thonhauser, P. Hyldgaard, and B. I. Lundqvist, Rep. Prog. Phys. 78, 066501 (2015).
- [38] K. Berland and P. Hyldgaard, Phys. Rev. B 89, 035412 (2014).
- [39] J. Klimeš, D. R. Bowler, and A. Michaelides, Phys. Rev. B 83, 195131 (2011).
- [40] T. Björkman, J. Chem. Phys. 141, 074708 (2014).
- [41] W. Li, N. Mingo, L. Lindsay, D. A. Broido, D. A. Stewart, and N. A. Katcho, Phys. Rev. B 85, 195436 (2012).
- [42] W. Li, L. Lindsay, D. A. Broido, D. A. Stewart, and N. Mingo, Phys. Rev. B 86, 174307 (2012).
- [43] A. Togo, F. Oba, and I. Tanaka, Phys. Rev. B 78, 134106 (2008).
- [44] O. Hellman and I. A. Abrikosov, Phys. Rev. B 88, 144301 (2013).
- [45] F. Eriksson, E. Fransson, and P. Erhart, Adv. Theory Simul. 2, 1800184 (2019).
- [46] L. Gharaee, P. Erhart, and P. Hyldgaard, Phys. Rev. B 95, 085147 (2017).
- [47] D. O. Lindroth and P. Erhart, Phys. Rev. B 94, 115205 (2016).
- [48] H. Anno, H. Yamada, T. Nakabayashi, M. Hokazono, and R. Shirataki, J. Solid State Chem. 193, 94 (2012).
- [49] N. Tsujii, J. H. Roudebush, A. Zevalkink, C. A. Cox-Uvarov, G. J. Snyder, and S. M. Kauzlarich, J. Solid State Chem. 184, 1293 (2011).
- [50] Y. Nagatomo, N. Mugita, Y. Nakakohara, M. Saisho, M. Tajiri, R. Teranishi, and S. Munetoh, J. Phys.: Conf. Ser. 379, 012008 (2012).
- [51] M. Christensen and B. B. Iversen, Chem. Mater. 19, 4896 (2007).
- [52] N. L. Okamoto, K. Kishida, K. Tanaka, and H. Inui, J. Appl. Phys. **100**, 073504 (2006).
- [53] S. Y. Rodriguez, L. Saribaev, and J. H. Ross, Phys. Rev. B 82, 064111 (2010).
- [54] Y. Takasu, T. Hasegawa, N. Ogita, M. Udagawa, M. A. Avila, K. Suekuni, I. Ishii, T. Suzuki, and T. Takabatake, Phys. Rev. B 74, 174303 (2006).
- [55] Y. Takasu, T. Hasegawa, N. Ogita, M. Udagawa, M. A. Avila, K. Suekuni, and T. Takabatake, J. Phys. Soc. Jpn. 77, 254 (2008).
- [56] Y. Takasu, T. Hasegawa, N. Ogita, M. Udagawa, M. A. Avila, K. Suekuni, and T. Takabatake, Phys. Rev. Lett. 100, 165503 (2008).
- [57] Y. Takasu, T. Hasegawa, N. Ogita, M. Udagawa, M. A. Avila, K. Suekuni, and T. Takabatake, Phys. Rev. B 82, 134302 (2010).
- [58] T. Mori, S. Goshima, K. Iwamoto, S. Kushibiki, H. Matsumoto, N. Toyota, K. Suekuni, M. A. Avila, T. Takabatake, T. Hasegawa, N. Ogita, and M. Udagawa, Phys. Rev. B 79, 212301 (2009).
- [59] K. Iwamoto, T. Mori, S. Kushibiki, H. Honda, H. Matsumoto, K. Suekuni, M. Avila, T. Takabatake, and N. Toyota, in *Materials Integration*, Key Engineering Materials Vol. 508 (Trans Tech Publications Ltd, Durnten-Zurich, Switzerland, 2012), pp. 341–346.
- [60] K. Iwamoto, S. Kushibiki, H. Honda, S. Kajitani, T. Mori, H. Matsumoto, N. Toyota, K. Suekuni, M. A. Avila, and T. Takabatake, J. Phys. Soc. Jpn. 82, 024601 (2013).
- [61] We note that since only certain vibrational motions can be detected with Raman and THz spectroscopy techniques, it is possible to indirectly draw some conclusions regarding the symmetries of the modes. Specifically, those that are Raman active have either T_{2g} or E_g symmetries and represent vibrations

parallel to the [100] and [110] directions [57]. With THz spectroscopy, however, the only visible, guest atom modes are those that are infrared active, and have T_{1u} symmetries [58]. Presumably, the lower and higher of these two modes correspond to motions perpendicular and parallel to the [001] direction (the out-of-plane direction in Fig. 1), respectively.

- [62] D. Nataraj and J. Nagao, J. Solid State Chem. 177, 1905 (2004).
- [63] M. A. Avila, K. Suekuni, K. Umeo, H. Fukuoka, S. Yamanaka, and T. Takabatake, Phys. Rev. B 74, 125109 (2006).
- [64] M. Avila, K. Suekuni, K. Umeo, and T. Takabatake, Physica B (Amsterdam) 383, 124 (2006).
- [65] J. Martin, H. Wang, and G. S. Nolas, Appl. Phys. Lett. 92, 222110 (2008).
- [66] A. F. May, E. S. Toberer, A. Saramat, and G. J. Snyder, Phys. Rev. B 80, 125205 (2009).
- [67] J. Tang, R. Kumashiro, J. Ju, Z. Li, M. A. Avila, K. Suekuni, T. Takabatake, F. Guo, K. Kobayashi, and K. Tanigaki, Chem. Phys. Lett. 472, 60 (2009).
- [68] M. Troppenz, S. Rigamonti, and C. Draxl, Chem. Mater. 29, 2414 (2017).
- [69] T. Uemura, K. Akai, K. Koga, T. Tanaka, H. Kurisu, S. Yamamoto, K. Kishimoto, T. Koyanagi, and M. Matsuura, J. Appl. Phys. **104**, 013702 (2008).
- [70] The "center" for each of the $3 \times 54 = 162$ phonon modes, in the form of the eigenvalues of the Hessian matrix, was taken as the atom that gave the largest contribution to the phonon density of states at that particular frequency.
- [71] A. Bentien, M. Christensen, J. D. Bryan, A. Sanchez, S. Paschen, F. Steglich, G. D. Stucky, and B. B. Iversen, Phys. Rev. B 69, 045107 (2004).
- [72] M. Christensen, S. Johnsen, M. Söndergaard, J. Overgaard, H. Birkedal, and B. B. Iversen, Chem. Mater. 21, 122 (2009).
- [73] X. Hou, Y. Zhou, L. Wang, W. Zhang, W. Zhang, and L. Chen, J. Alloys Compd. 482, 544 (2009).

- [74] Y. Liu, Q. Xi, J. Zhou, T. Nakayama, and B. Li, Phys. Rev. B 93, 214305 (2016).
- [75] Q. Xi, Z. Zhang, J. Chen, J. Zhou, T. Nakayama, and B. Li, Phys. Rev. B 96, 064306 (2017).
- [76] Q. Xi, Z. Zhang, T. Nakayama, J. Chen, J. Zhou, and B. Li, Phys. Rev. B 97, 224308 (2018).
- [77] E. Fransson, F. Eriksson, and P. Erhart, arXiv:1902.01271.
- [78] We do not include the data from Ref. [24] in Fig. 5 since it is based on a different exchange-correlation functional (PBE) and a different model for the distribution of Ga and Ge over the host lattice sites, which would lead to a misleading comparison.
- [79] See Supplemental Material at http://link.aps.org/supplemental/ 10.1103/PhysRevB.100.045206 for additional analysis of lifetimes and thermal conductivity, as well as an overview of how experimental studies of the thermal conductivity handled the electronic contribution to the thermal conductivity.
- [80] G. K. H. Madsen, A. Katre, and C. Bera, Phys. Status Solidi A 213, 802 (2016).
- [81] J.-Y. Yang, L. Cheng, and M. Hu, Appl. Phys. Lett. 111, 242101 (2017).
- [82] P.-F. Lory, S. Pailhès, V. M. Giordano, H. Euchner, H. D. Nguyen, R. Ramlau, H. Borrmann, M. Schmidt, M. Baitinger, M. Ikeda, P. Tomeš, M. Mihalkovič, C. Allio, M. R. Johnson, H. Schober, Y. Sidis, F. Bourdarot, L. P. Regnault, J. Ollivier, S. Paschen, Y. Grin, and M. de Boissieu, Nat. Commun. 8, 491 (2017).
- [83] H. Euchner, S. Pailhès, V. M. Giordano, and M. de Boissieu, Phys. Rev. B 97, 014304 (2018).
- [84] K. Biswas, J. He, I. D. Blum, C.-I. Wu, T. P. Hogan, D. N. Seidman, V. P. Dravid, and M. G. Kanatzidis, Nature (London) 489, 414 (2012).
- [85] G. D. Mahan and M. Bartkowiak, Appl. Phys. Lett. 74, 953 (1999).
- [86] T. J. Scheidemantel, C. Ambrosch-Draxl, T. Thonhauser, J. V. Badding, and J. O. Sofo, Phys. Rev. B 68, 125210 (2003).